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_____ is a representation of a chemical reaction with reactants on the left, products on the right, and an arrow separating the two.

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precipitate. bubbles. signs of a chemical reaction. subscript. Solid compound produced from 2 aqueous solutions during a chem.... Gas given off during a chemical reaction; one of the signs of....

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change in heat/light, formation of a gas (bubbles), solid prec....

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separating the two. Small whole number ratios that are placed in front of the formulas in an equation in order to balance it.

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negative number that is assigned to an atom to indicate its degree of oxidation or reduction. The term oxidation state is often used interchangeably with oxidation number.

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Chemical Equation n Describes chemical reaction. n Chemical equation: reactants yield products n Reactants à Products n Much easier to write symbols and formulas instead of words.

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Section 11.1 – Describing Chemical

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Reactions In a chemical reaction, the reactants are written on the left and the products on the right. The arrow that separates them is called yield. Reactants
□ Products

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6th period . 1) Combination Reactions •
Is also referred to as a synthesis

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reaction • It is a chemical change in which two or more

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Question 1. Give an example for
complex chemical reaction. Answer:

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'Photo synthesis' is a complex chemical reaction. Question 2. Give an example of simple for chemical reactions. Answer: Rusting of iron and combustion of fuels are simple chemical reactions. Question 3.

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Chemistry Chapter 11 Chemical Reactions a representation of the chemical reaction; the formulas of the.... a chemical equation that does not indicate the relative amount.... a substance that speeds up a reaction but is not used up in th.... small whole numbers placed in front of the formulas in the equ.... chemistry chapter 11

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either releases or absorbs thermal energy by chemical processes. Answer key to chapter 11 chemical reactions

Chapter 11 Chemical Reactions Test Answer Key

The process of breaking food down in the body in order to absorb what is needed and more easily

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eliminate - this involves chemical reactions. Define: Chemical Equation: A "chemical equation" is a representation of a chemical reaction; the formulas of the "reactants" are connected by an "arrow (->)" with the formulas of the "product".

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Types of Chemical Reactions > 13 A

decomposition reaction is a chemical change in which a single compound breaks down into two or more simpler products. • Decomposition reactions involve only one reactant and two or

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more products.

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oxidation-reduction reactions A chemical
reaction that exhibits a change in the

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oxidation states of one or more elements in the reactants that has the general form oxidant + reductant \rightarrow reduced oxidant + oxidized reductant. The general forms of these five kinds of reactions are summarized in Table 11.6.1, along with examples of each.

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